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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom's

### **5 The Periodic Law**

Modern Chemistry Chapter 5. periodic law. lanthanide. actinide. alkali metals. The physical and chemical properties of the elements are perio.... The 14 elements with the atomic numbers 58-71. The 14 elements with the atomic numbers 90-103. The elements of group 1 of the periodic table.

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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 \_\_\_\_ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. \_\_\_\_ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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