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Lecture Notes 0.79 Mnitrogen+ 0.21 Moxxygen. =
 $0.79(28) + 0.21(32) = 28.84$ The mass
fractions of oxygen and nitrogen in air are

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then $m_{\text{oxygen}} = (0.21)(32)/28.84 = 0.233$, or 23.3% and $m_{\text{nitrogen}} =$

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4. Write additional data required to solve the problem and the chemical equations if the process involves chemical reaction. 5. Select a suitable basis of calculations. 6. List by symbols each of the unknown values of the stream flows and compositions 7. Make a number of independent material balances equations equal to unknown

Basic Principles and Calculations in Chemical Engineering

Lec 1: Definition, History, Role of Chemical

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Engineer; Lec 2: Basic Features of Chemical Process; Lec 3: Unit systems and dimensions; Process Variables and Rate. Lec 4: Variables and Properties of Material in System; Lec 5: Pressure and Temperature of Flow Process; Lec 6: Rate of Process; Fundamentals of Material Balance

NPTEL :: Chemical Engineering - NOC:Basic Principles and ...

view the study of the field of chemical engineering as a tree with material and energy balances being the trunk and the subjects of thermodynamics, fluid flow, heat

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transfer, mass transfer, reactor kinetics, process control, and process design being the branches off the trunk. From this perspective, it is

Basic Principles and Calculations in Chemical Engineering

$z + z_P$. $z = \hat{g}$ (2.2.8) Setting the limit on the left hand side as $z \rightarrow 0$ gives the definition of the derivative of P with respect to z : $\frac{dP}{dz} = -\hat{g}$ (2.2.9) In other words, the change in pressure, P , with height, z , is negative - pressure reduces with height (or increases with depth - as any diver knows).

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MIT course 10.34 Numerical Methods Applied to
Chemical Engineering of Fall 2015, taught by

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Prof. William Green, Jr. and Prof. James W.
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