

Chapter 6 Thermochemistry Test

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Chapter 6 Thermochemistry 11 Chapter 6 Thermochemistry Energy Flow and Chemical Change part 1 Chapter 6 thermochemistry 1 34 Chapter 6 Thermochemistry Review Chem-1a AP Chem- Thermochem-4 Ch-6 Zumdahl Chapter 6 - Thermochemistry - 10/19 ~~Chapter 6: Thermochemistry (37:23)~~ Chapter 6 Thermochemistry Test
162 CHAPTER 6: THERMOCHEMISTRY To convert the answer to joules, we write: 101.3 J 0.18 L atm 1L atm = - x = w - 18 J 6.17 An expansion implies an increase in volume, therefore w must be -325 J (see the defining equation for pressure-volume work.)

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164 CHAPTER 6: THERMOCHEMISTRY Substituting into the above equation: E = 483.6 × 103 J - (8.314 J/mol K)(398 K)(+1 mol) E = 4.80 × 105 J = 4.80 × 102 kJ 6.28 We initially have 6 moles of gas (3 moles of chlorine and 3 moles of hydrogen). Since our product is 6

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Ch.6 - Thermochemistry Ch.6.1: The Nature of Energy Energy: An object 's capacity to perform work or produce heat Potential Energy: Energy due to position or composition (chemical bonds). Kinetic Energy: Energy due to the motion of the object 1 2 2 KE mv Law of Conservation of Energy: Energy can neither be created nor destroyed.

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