

Chapter 6 Chemistry Review

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between the nuclei and valence electron of different atoms
that binds the atoms together.

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Chapter 6 Chemistry Review Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank.

____ 1. The electrons involved in the formation of a chemical bond are called a. dipoles. c. Lewis electrons. b. s electrons. d. valence electrons. ____ 2.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT

ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

6 Chemical Bonding

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molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a. molar mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg SO_3 1 g 1000 mg 1 mol 80.07 g = 6.14 10-4 mol SO_3 b. molar mass of $PbO_2 = 239.2 \text{ g}$ 7.44 x 104 kg PbO_2 1000 g 1 kg 1 mol 239.2 g = 3.11 105 mol PbO_2 c. molar mass of $CHCl_3 = 119.37 \text{ g}$ 59.1 g $CHCl_3$ 1 mol 119.37 g = 0.495 mol $CHCl_3$ d. molar mass of $C_2H_3Cl_3 = 133.39 \text{ g}$ 3.27 ...

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answerhalogen atom bonded to an sp^3 hybridized carbon
atom that is tetrahedral questionprimary alkyl halide ...

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Bond. Chapter 6 - Answer Key to Section Review 1-3. Section
Review 1 1. What is the main distinction between ionic and
covalent bonding? Answer (A): Ionic bonding involves the
electrical attraction between large numbers of anions and
cations. Covalent bonding involves the sharing of electron
pairs... <https://www.scribd.com/document/173653258/Chemistry-Chapter-6-Review>

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6: An Overview of Organic Reactions. This chapter is
designed to provide a gentle introduction to the subject of
reaction mechanisms. Two types of reactions are
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thermodynamics and bond dissociation energies.

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